

# The fate of xenobiotics in the body

## Xenobiotic

- A substance that causes pathological changes in the body after absorption into the bloodstream
- in other words also noxious substance or poison
- Toxic effects:
  1. transient
  2. permanently damaging
  3. fatal

## The fate of a drug in the body

**Pharmacodynamics** - what the poison does to the organism (effects)

**Pharmacokinetics** - what the organism does with the poison (toxicological analyses)

- it includes following processes:

1. Administration - a way xenobiotic enters the body: Inhalation - snorting - injection - per os - dermal....

Potential release of the bound active ingredient in the drug

1. **Absorption** - the process of diffusion into the bloodstream
2. **Distribution** - between blood and tissues and vice versa
3. **Biotransformation** - a transformation of chemical structure, formation of active and inactive metabolites
4. **Elimination** Routes of excretion - depend on a substance polarity: Kidneys - Intestines - Skin - Saliva - Lungs - Hair - Nails

## Absorption of a drug

- passive by diffusion or
- by active transport

**Speed of absorbance and absorbance rate - route of administration**

1. Intravenous - rapid and complete absorption (100%)
2. Pulmonary inhalation - rapid and reduced absorption
3. Parenteral administration - absorption from tissues - circulation
4. Oral - first pass effect

**Extent of xenobiotics absorption in the digestive tract**

1. Environment pH influences the drug absorption
2. Absorption according to the acid-base properties of the drug

<b>Stomach</b>	pH 1-3	pole B1	< 5 m <sup>2</sup>	acidic substances (aspirin)
<b>Small intestine S<sub>1</sub></b>	pH 5-7	pole B2	200 m <sup>2</sup>	
<b>Small intestine S<sub>2</sub></b>	pH 7-8	pole B3	< 5 m <sup>2</sup>	
<b>Colon</b>	pH 7-8	pole B3	< 5 m <sup>2</sup>	base (ephedrine)

**The absorbed proportion of the drug - depends on the method of administration**

- Biological availability (Bioavailability)

Proportion (%) of the drug absorbed into the bloodstream by a certain administration route in infinite time relative to i.v. administration (100%)

- „First pass metabolism“ - „First pass effect“
  - Proportion of the drug that reaches the liver before entering the circulation, and therefore is metabolized before it expresses its pharmacological effect - reduction in bioavailability
  - Pre-systemic metabolite formation in oral administration compared to subcutaneous administration (sc.)

## Distribution

Model of the body as a set of compartments

**Distribution depends on**

- The polarity and size of the drug molecule
- Binding of the drug and its metabolites to plasma proteins
- Extent of ionization at given plasma pH
- Blood supply to tissues, transport of the drug
- Distribution between blood and tissues - water content of tissues

*e.g.*

- *Alcohol is dissolved evenly in the water throughout the body*
- *Warfarin is strongly bound to plasma proteins and is only found in plasma and extracellular fluid*

TISSUE	% of water
muscle	76
fat	10
liver	68
kidneys	83
lungs	79
brain	75

	Volume of water /body mass (l/kg)
<b>plasma</b>	0,04
<b>blood</b>	0,08
<b>extracellular water</b>	0,2
<b>total body water</b>	0,6
<b>fat</b>	0,2-0,35

- The volume of total water decreases in course of individual's development.
- **Distribution volume** - hypothetical quantity after distribution equilibrium is reached. Substances strongly bound to tissue proteins have high Vd and reduced plasma concentration

$V_d = D/c_0$  nebo  $V_d = a/c$   
 D – absorbed dose of a drug/chemical (i. v. dose)  
 $c_0$  – initial plasma concentration (after i. v. dose)

(see reading from the semi-logarithmic kinetic curve below)

a – the current amount of the drug in the body  
 c – current plasma concentration

Distribution volume of substances

<b>V<sub>d</sub> (l)</b>	<b>Place of occurrence of the drug</b>
<b>5</b>	remains in plasma
<b>5-20</b>	distribution into the extracellular space
<b>20-40</b>	distribution into all body fluids
<b>&gt;40</b>	depot in peripheral tissues

## Elimination

The vast majority of xenobiotics is metabolized in the liver and excreted in the urine. Ethanol is eliminated according to zero- order kinetics. First-order kinetics mainly applies to drugs:

–  $dc/dt = k_{el} \cdot c$

Plasma elimination half-life:

–  $dc/dt = k_{el} \cdot c$   
 $c = \exp(-k_{el} \cdot dt)$   
 $\ln c = -k_{el} \cdot t$

Half-life when,  $c = \frac{1}{2} c_0$

$\ln \frac{1}{2} = -k_{el} \cdot t_{\frac{1}{2}}$

$$\ln 2 = k_{el} \cdot t_{1/2}$$

$$t_{1/2} = 0,693 / k_{el}$$

- In general: after 5 half-lives, 96.875% of the drug is eliminated, i.e. the organism is practically clean of the substance
- Clearance – volume of plasma from which a substance is completely removed per unit time

$$[CL] = [ \text{volume} / (\text{mass} \cdot \text{time}) ]$$

- Total clearance and clearance of individual organs - dependent on the health condition of the individual
- Substances bound in tissues with high  $V_d$  are eliminated for a long time at a given clearance, they have a long half-life

$$t_{1/2} = 0,693 \cdot V_d / CL \text{ or } k_{el} = CL / V_d$$

- SUMMARY - Important pharmacokinetic data:

$$k_{el}; t_{1/2}; V_d; CL$$

## Enterohepatic circulation

- Xenobiotics and metabolites excreted by the bile into the intestines are reabsorbed, then they pass the liver and are again excreted into the bile, then again into the intestines.....
- This cycle is repeated until complete elimination
- Circulation prolongs the stay of a xenobiotic in the organism and can prolong or delay toxic manifestations
- Chemicals which are proved to enter enterohepatic circulation are found in the intestines even after parenteral administration (e.g. opiates, benzodiazepines)

## Release from tissues

- Tissue blood supply
- Adipose tissue - little blood supply
- Slow re-release of an accumulated drug into the blood
- Extended elimination

*e.g. Chronic marijuana abuse - depot in adipose tissue - long-term metabolite excretion*

## Links

## Related articles

- Toxicity, effects of noxious substances
- Introduction to toxicology
- Pharmacodynamics
- Pharmacokinetics

## Source

- BALÍKOVÁ, Marie. *Osud xenobiotik a biotransformace* [online]. [cit. 2012-03-13]. <<https://el.lf1.cuni.cz/p88978866/>>.
- PROKEŠ, Jaroslav. *Základy toxikologie : obecná toxikologie a ekotoxikologie*. 1. edition. Prague : Galén : Karolinum, c2005. ISBN 80-7262-30-1X.