

Isoosmolal

A solution whose real osmolality is the same as the real osmolality of another solution is called **isoosmolal**. In the organism, the osmolality of solutions is most often compared with the real osmolality of plasma, i.e. solutions that have the same real osmolality as plasma are referred to as iso-osmolal. This is, for example, a 0.9% NaCl solution or a 5% glucose solution.

Links

Source

- ŠVÍGLEROVÁ, Jitka. *Izoosmolální* [online]. The last revision 18. 2. 2009, [cit. 11.11.2010]. <<https://web.archive.org/web/20160306065550/http://wiki.lfp-studium.cz/index.php/Izoosmolální>>.