

Cysteine

Cysteine is a non-essential amino acid that, like methionine, has a **sulfur** atom in its molecule.

Metabolism cysteine

Cysteine synthesis takes place in the human organism from **homocysteine** and **serine**.

The degradation of cysteine begins with the oxidation of the -SH group to $-SO_2^-$ by the enzyme cysteine dioxygenase. In the resulting **L-cysteine sulfinic acid**, the **-NH₂** group is replaced by a keto group with the help of transaminase, and **β -sulfanylpyruvate** is formed. In the final reaction, it is split by desulfinase into **pyruvate** and sulfite (SO_3^-), or final sulfate (SO_4^-). Cysteine is an important source of **taurine**. L-cysteine sulfinic acid is decarboxylated to **hypotaurine** and subsequent oxidation of the $-SO_2^-$ group to $-SO_3^-$ produces taurine.

Alternative non-oxidative degradation of cysteine produces **pyruvate** and **sulfane** (H_2S).

Importance

- In peptides, cysteine is essential for the formation of **disulfide bridges**.
- It is a substrate for **glutathione**.
- It is a substrate for **taurine**. The latter is conjugated with bile acids or other substances that increase their solubility in water.
- Decarboxylation of cysteine produces **cysteamine**, which is part of **coenzyme A**.
- It has a high proportion of **keratin** protein (hair, nails).

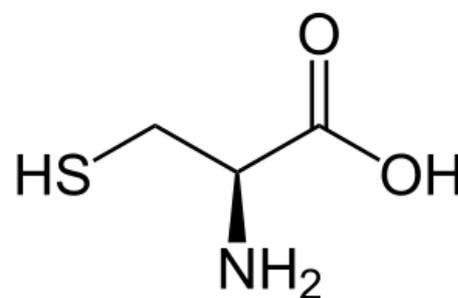
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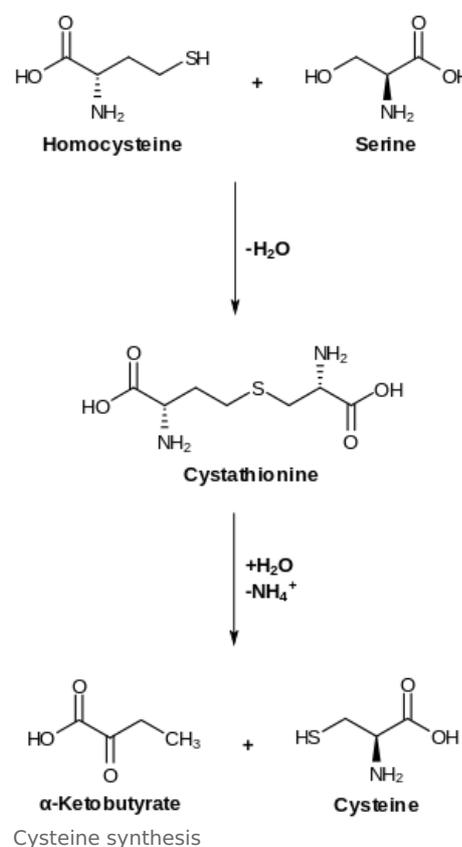
Aminoacids

References

- MATOUŠ, Bohuslav, et al. *Fundamentals of medical chemistry and biochemistry*. 2010 edition. Prague: Galen, 2010. 0 pp. ISBN 978-80-7262-702-8.



Cysteine molecule



Cysteine synthesis