

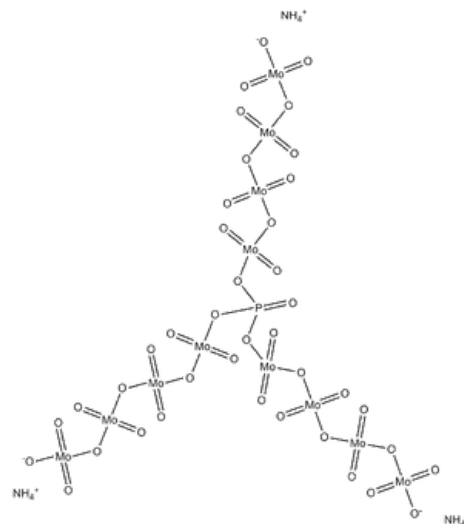
Ammonium phosphomolybdate

Ammonium phosphomolybdate, $(\text{NH}_4)_3\text{PMo}_{12}\text{O}_{40}$ is a yellow, poorly soluble substance. It is formed by the reaction of phosphates and ammonium molybdate under heat; of this reaction is used for the detection of phosphates in solutions.

Links

External links

- Ammonium phosphomolybdate (wikipedia)
- Chemicalbook.com article on ammonium phosphomolybdate ([https://](https://www.chemicalbook.com/ChemicalProductProperty_EN_CB6162083.htm)



Structural formula of ammonium phosphomolybdate

www.chemicalbook.com/ChemicalProductProperty_EN_CB6162083.htm)